UNIVERSITY OF BELGRADE TECHNICAL FACULTY IN BOR

BOOK OF ABSTRACTS



8th INTERNATIONAL STUDENT CONFERENCE ON TECHNICAL SCIENCES



NWW. Tibor. bg. ac. rs



20-21 October, Bor Lake, Serbia

Editor: Uroš Stamenković



Book of Abstracts,

8th International Student Conference on Technical Sciences ISC 2023

Editor:

Doc. dr Uroš Stamenković

University of Belgrade - Technical Faculty in Bor

Technical Editors: Milan Nedeljković, dipl. ing. Avram Kovačević, dipl. ing.

University of Belgrade - Technical Faculty in Bor

Publisher: University of Belgrade - Technical Faculty in Bor

For the publisher: Dean, Prof. dr Dejan Tanikić

Circulation: 50 copies Year of publication: 2023

Printed by "GRAFIKA GALEB DOO" NIŠ, 2023

ISBN 978-86-6305-141-6

СІР - Каталогизација у публикацији Народна библиотека Србије, Београд

622(048) 669(048) 66(048) 66.017/.018(048)

INTERNATIONAL Student Conference on Technical Sciences (8; 2023; Borsko jezero)

Book of abstracts / 8th International Student Conference on Technical Sciences ISC 2023, 20-21 October, Bor Lake, Serbia; [organized by University of Belgrade, Technical Faculty in Bor]; editor Uroš Stamenković. - Bor: University of Belgrade, Technical Faculty, 2023 (Niš: Grafika Galeb). - VII, 51 str.; 24 cm

Tiraž 50. - Bibliografija uz većinu apstrakata.

ISBN 978-86-6305-141-6

а) Рударство -- Апстракти b) Металургија -- Апстракти v) Хемијска технологија -- Апстракти g) Технички материјали -- Апстракти

COBISS.SR-ID 126594825



October 20th – 21st, 2023, Bor lake in Bor (Serbia) www.tfbor.bg.ac.rs https://ioc.tfbor.bg.ac.rs/isc2023/

8th International Student Conference on Technical Science, ISC 2023.

Is organized by

UNIVERSITY OF BELGRADE, TECHNICAL FACULTY IN BOR

and co-organized by

University of Zenica, Faculty of engineering and natural sciences, Zenica, Bosnia and Herzegovina

University in Priština, Faculty of Technical Science, Kosovska Mitrovica, Serbia:

University of Montenegro, Faculty of Metallurgy and Technology, Podgorica, Montenegro;

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TRACKING THE ABSORPTION ABILITY OF EXHAUST GASES MODEL MIXTURE USING AN AQUEOUS SOLUTIONS OF NaOH AND KOH

Students: Željka Nikolić, Adrijana Šutulović, Boris Rajčić, Dubravka Milovanović, Vladimir Nikolić, Zoran Šaponjić

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Abstract

Introduction and objective

Emission from vehicles is one of anthropogenic sources of air pollution [1]. Major pollutants emitted from fossil-fuel internal combustion engine are carbon monoxide (CO), carbon dioxide (CO₂), oxides of nitrogen (NOx), particulate matter (PM) and hydrocarbons (HC) including volatile organic compounds (VOCs) and polycyclic aromatic hydrocarbons (PAHs) [2,3]. The aim of this probe was to examine the absorption ability of exhaust gases model mixture with an aqueous solutions of sodium hydroxide (NaOH) and potassium hydroxide (KOH).

Methodology

Components of exhaust gases model mixture were CO, CO₂, methane, ethene and ethane. Gaseous mixture was introduced in aqueous solutions of NaOH and KOH through glass diffuser [4,5]. During the introduction of gases turbidity occurrence and the change in alkalinity were monitored. Changes in gas concentrations by passing through the hydrokside solution were measured by separation technique Gas Chromatography (GC) with Thermal Conductivity Detector (TCD) and Flame Ionization Detector (FID).

Results

During the introduction of gaseous mixture in aqueous solutions of NaOH and KOH turbidity and precipitation occur in proportion to the time. There was no measurable change in alkalinity of solutions in time for 160 min. The signal from CO_2 is descended below measurable value while signals from CO, CH_4 , C_2H_4 and C_2H_6 shown the same value prior to absorption.

Conclusions

Difference in composition of exhaust gases model mixture indicates that an aqueous solutions of NaOH and KOH completely absorbs CO₂ until complete saturation. However, compounds CO, CH₄, C₂H₄ and C₂H₆ were not absorbed at all with these solutions.

Keywords: Exhaust gases, Gaseous pollutants, Air quality, Environment protection, Fossi-fuels

ACKNOWLEDGEMENT

The Ministry of Science, Technological Development and Innovation of Republic of Serbia supported this study (Contract number: 451-03-47/2023-01/200051).



October $20^{\rm th}-21^{\rm st},\,2023,\,$ Bor lake in Bor (Serbia) www.tfbor.bg.ac.rs https://ioc.tfbor.bg.ac.rs/isc2023/

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